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Stoichiometry vocab. stoichiometry. mole ratio. excess reactant. limiting reactant. The study of quantitative relationships between the amounts of.... In a balanced equation, the ratio between the numbers of moles.... A reactant that remains after a chemical reaction stops.

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combination reaction. decomposition
reaction. single-replacement reaction.
activity series. a chemical change in
which two or more substances react to

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for.... a chemical change in which a single compound is broken down in.... a chemical change in which one element replaces a second eleme....

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barometer. standard atmospheric

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pressure. manometer. daltons law of partial pressures. a device that measures atmospheric pressure. It is filled with.... 1 atm = 760 mm Hg...
pascal = 1N/m^2 ... 1 atm = 101.3...
kpa = 101,325 Pa. measures gases in a closed container.

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convert mass/mole of one substance to another using mole fraction.

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movement, saturated, unsaturated, supersaturated, miscible, immiscible,

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Reduction is a chemical change that
involves the _____ of electrons. oxidation.
A(n) _____ number is the number of
electrons that are added or

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propane. -The formulas of organic compounds are written with carbon fir....
1. 2. 3.

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chemical reaction' the formulas o.... a
chemical equation that does not indicate
the relative amount.... a substance that
increases the rate of reaction by
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barometer. standard atmospheric pressure. manometer. dalton's law of partial pressures. a device that measures atmospheric pressure. It is filled with.... $1 \text{ atm} = 760 \text{ mm Hg}$...
 $\text{pascal} = 1 \text{ N/m}^2$... $1 \text{ atm} = 101.3$...
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