

Chapter 18 Chemistry Solutions

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18.1: Sketch a galvanic cell, and explain how it works. Look at Figs. 18.... 18.2: In making a specific galvanic cell, explain how one decides on the ... 18.3: You want to plate out nickel metal from a nickel nitrate solution o... 18.4: A copper penny can be dissolved in nitric acid but not in hydrochlo...

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Oxidizing agents (are reduced): H^+ ($E^\circ_{red} = 0V$). Reducing agents (are oxidized): Ni ($E^\circ_{ox} = + 0.23V$), Au ($E^\circ_{ox} = - 1.50V$), Mg ($E^\circ_{ox} = + 2.37V$). Both: Sn^{2+} + ($E^\circ_{red} = - 0.14V$; $E^\circ_{ox} = - 0.15V$), Fe^{2+} + ($E^\circ_{red} = - 0.44V$, $E^\circ_{ox} = - 0.77V$), Cl_2 ($E^\circ_{red} = 1.36V$, $E^\circ_{ox} = - 1.47V$).

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Solutions Manual Chemistry: Matter and Change • Chapter 18 371 CHAPTER18 SOLUTIONS MANUAL. 92. Air Purifier Lithium hydroxide is used to purify air by removing carbon dioxide. A 25.00-mL sample of lithium hydroxide solution is titrated to an endpoint by 15.22 mL of 0.3340Mhydrochloric acid solution.

Acids and BasesAcids and Bases - Weebly

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(b) the binding energy. - Homework Help and Textbook Solutions

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