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2 ion. Two F atoms each gain 1 e, forming 1 ions. The ions attract, forming SrF₂. 1 Sr ion (-2 Sr ion) 2 F ions (-1 F ion) $1(2) + 2(1) = 0$ The overall charge on one formula unit of SrF₂

Ionic Compounds and Metals

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Answer Key

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CHAPTER 7 REVIEW. Chemical Formulas and Chemical Compounds. SECTION 1. SHORT ANSWER Answer the following questions in the space provided. 1. cIn a Stock system name such as iron(III) sulfate, the Roman numeral tells us. (a)how many atoms of Fe are in one formula unit.

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threat assessment and management, which helps to ensure safe and continuous business operation and the availability, confidentiality, and integrity of an enterprise's information ...

Answers to Chapters

1,2,3,4,5,6,7,8,9 - End of Chapter ...

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