

Principles Of Descriptive Inorganic Chemistry

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BASIC PRINCIPLES OF INORGANIC CHEMISTRY

Inorganic chemistry is a broad and complex field. The underlying principles and theories are normally dealt with at a rather high level in a course that is normally taught at the senior level. With the emphasis on these topics, there is little time devoted to the descriptive chemistry of the elements.

Descriptive Inorganic Chemistry - Elsevier

These descriptive facts are taught to illustrate and illuminate fundamental chemistry principles. Elements Organized by Block. Elements Organized by Group. Elements Organized by Period. Periodic Trends of Elemental Properties. Main Group Reactions. Back to top. Stoichiometry and Balancing Reactions. Elements Organized by Block.

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Acknowledgments xix Dedication xxi CHAPTER 1 The Electronic
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Spectroscopy 2 1.1 The Schrödinger Wave Equation and Its
Significance 3 1.2 Shapes of the Atomic Orbitals 5 1.3 The
Polyelectronic Atom 9 1.4 Ion Electron Configurations 14

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Introduction to Modern Inorganic Chemistry begins by explaining the electronic structure and properties of atoms, then describes the principles of bonding in diatomic and polyatomic covalent molecules, the solid state, and solution chemistry.

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